

# Anomalous europium luminescence in $\text{LaF}_3$

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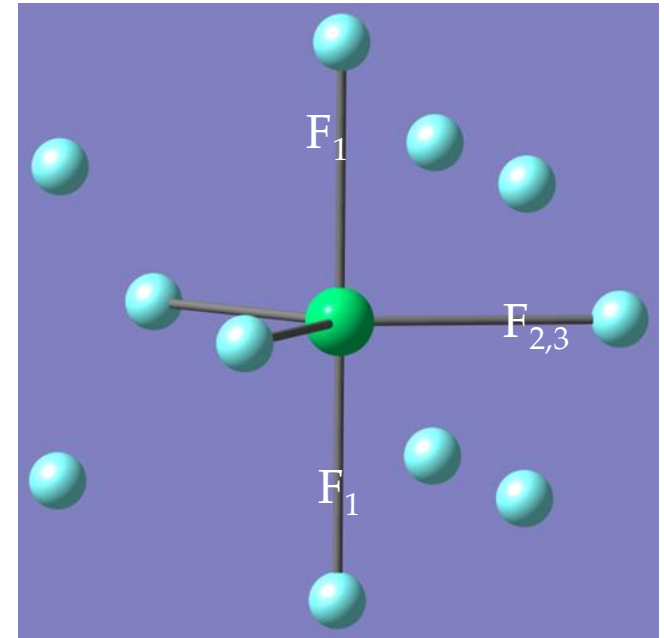
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# Introduction

Apart to detailed studies of  $\text{LaF}_3\text{-Re}^{3+}$  the spectroscopy of  $\text{Re}^{2+}$  in this host remains uninvestigated. Europium  $\text{Eu}^{2+}$  ions are known as very efficient luminescence impurity in dense scintillating hosts. Besides the normal 5d-4f luminescence in most materials, the  $\text{Eu}^{2+}$  shows so called anomalous luminescence with large Stokes shift near 1 eV in certain crystals. No data on absorption and luminescence of  $\text{Eu}^{2+}$  in  $\text{LaF}_3$  were found.

The aim of this paper is investigation of dielectric and optical spectra of divalent Eu, Sm in crystal  $\text{LaF}_3$  and clarification the nature of charge compensator as well as its influence on optical transitions.

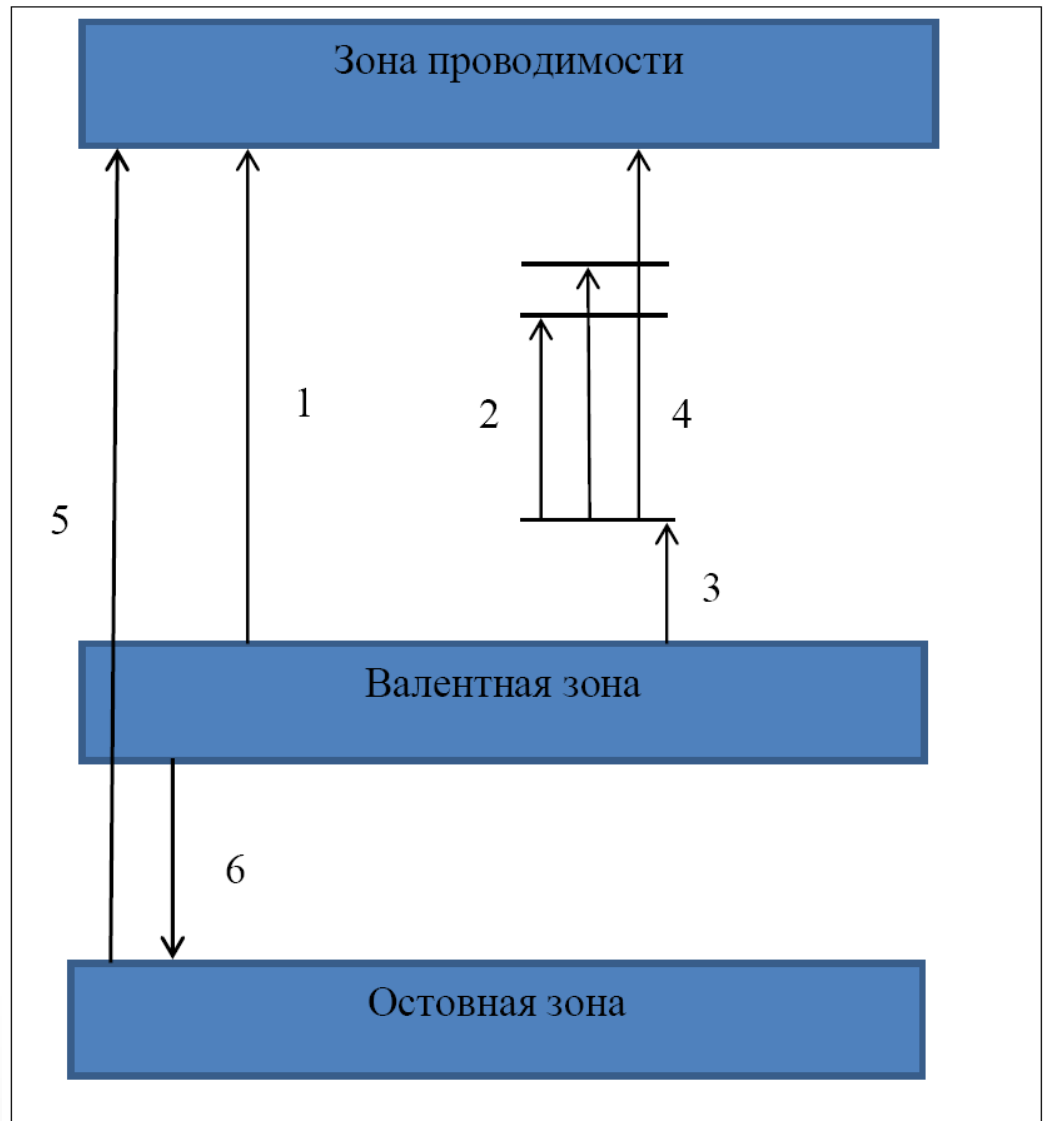


La in centre and nearest fluorines in  $\text{LaF}_3$  lattice  
 $D_{3h}$ -symmetry

# Типы переходов в диэлектриках

Основные типы переходов в диэлектрических кристаллах.

- 1 – межзонные переходы,
- 2 – внутрицентровые переходы,
- 3 – переходы с переносом заряда,
- 4 – фотоионизация,
- 5 – переходы с верхней основной зоны в зону проводимости,
- 6 – кросслюминесценция (основно-валентные переходы)



# Conductivity

A feature of  $\text{LaF}_3$  is high electrical conductivity undoped crystals at room temperature, which amounts to about  $10^{-6} \text{ ohm}^{-1}\text{cm}^{-1}$ . The conductivity of lanthanum fluoride for several orders of magnitude higher conductivity of alkaline earth fluorides. Numerous studies have found that the conductivity is due to the migration of  $\text{LaF}_3$  fluorine vacancies. When introducing divalent ion Ba, Sr, Ca the  $\text{LaF}_3$  conductivity increases significantly, that due to the presence of anionic vacancies.

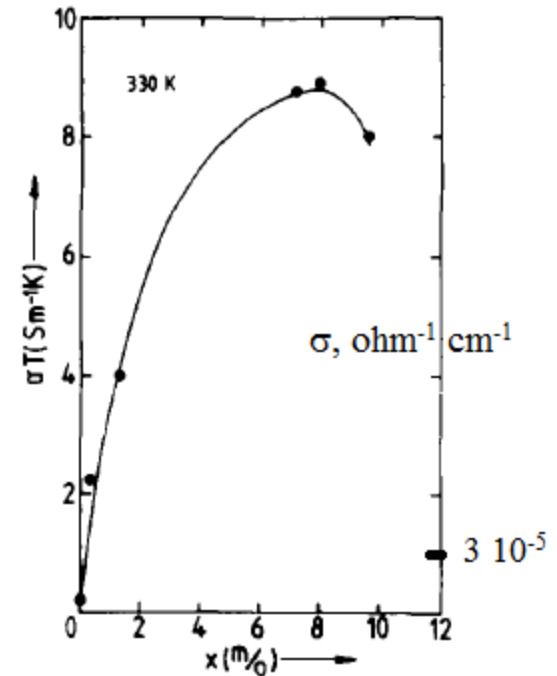
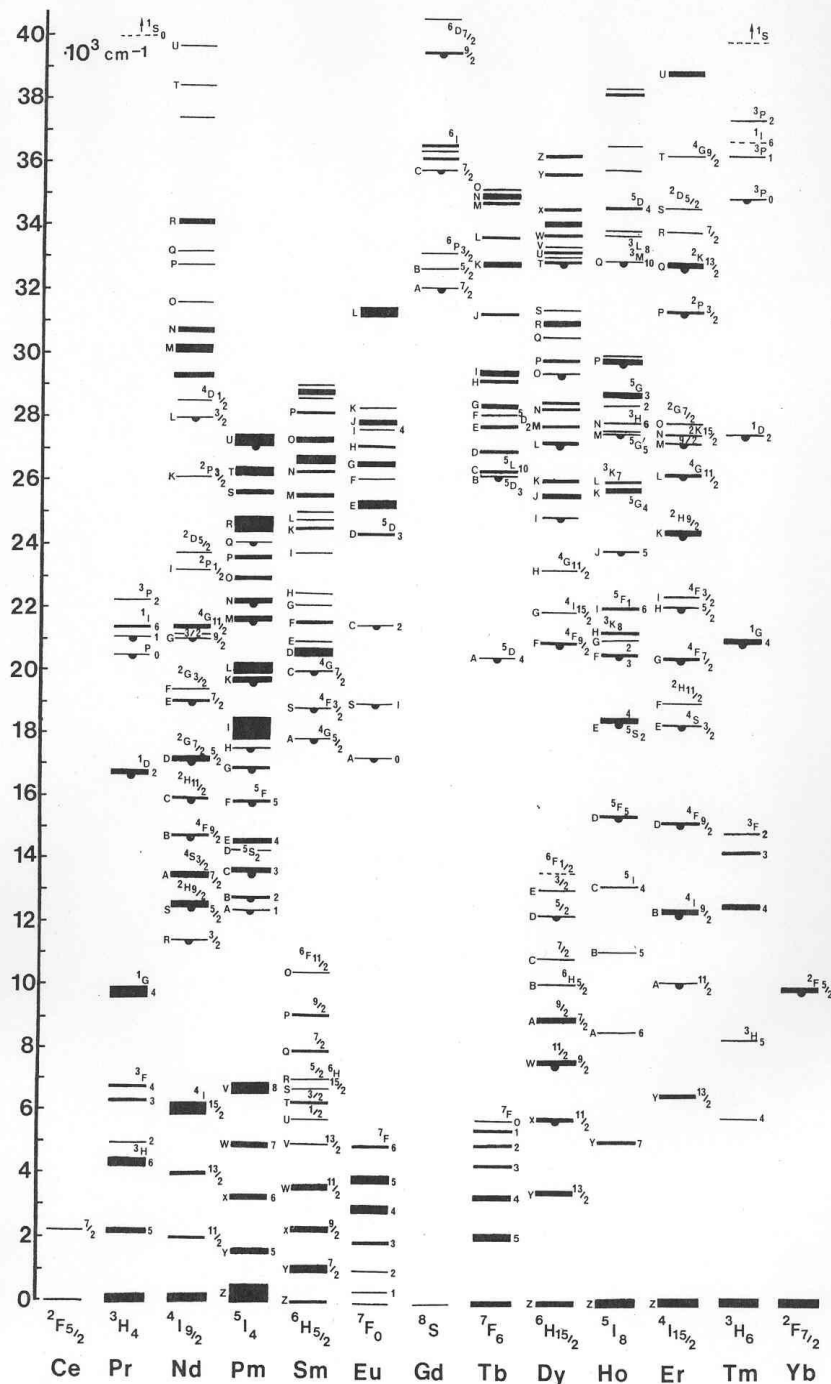


Fig. 3. The isothermal conductivity at 330 K plotted as  $\sigma T$  versus  $x$  for the solid solutions  $\text{La}_{1-x}\text{Ba}_x\text{F}_{3-x}$  ( $x$  axis).

A.Roos et al. Sol.St.Ionics, 13 (1984) 191-203

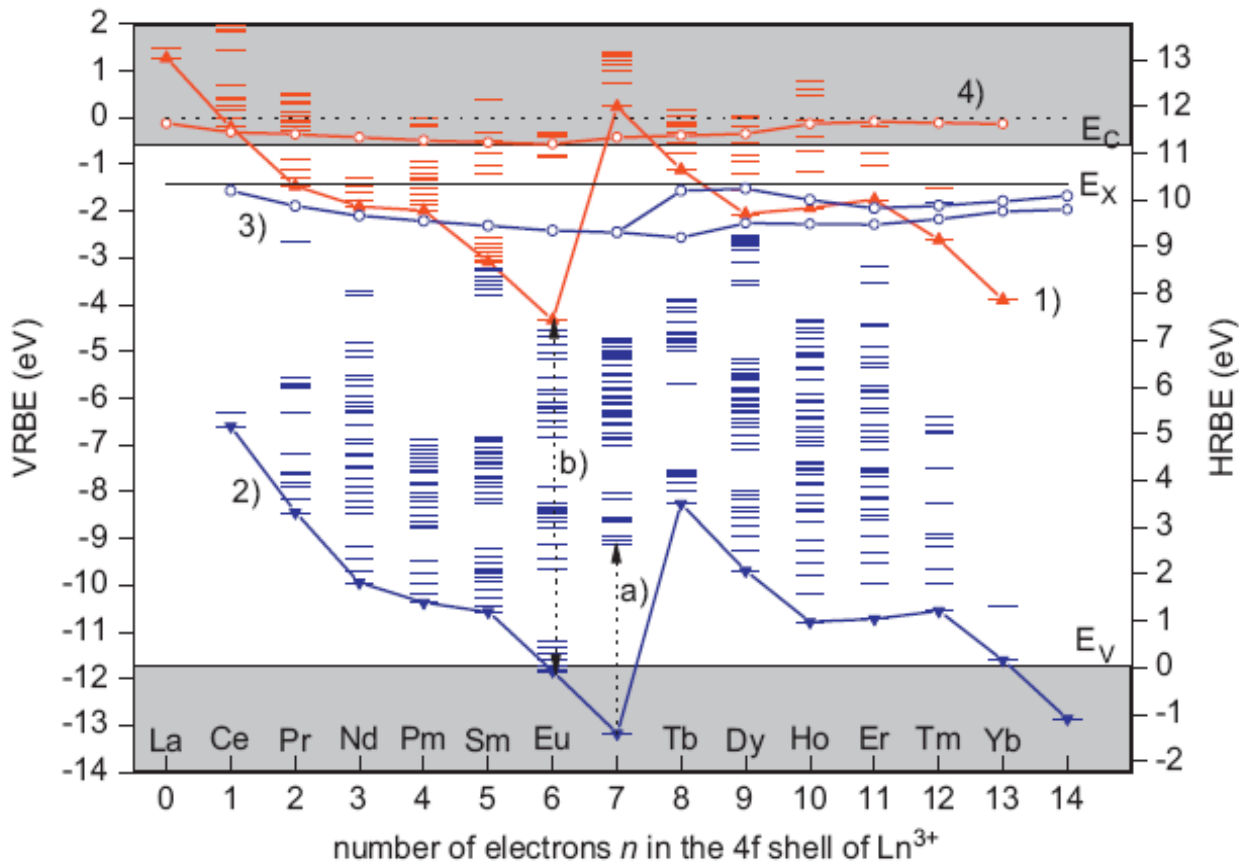
# Уровни f-оболочки

Диаграмма уровней энергии  
 трехвалентных лантаноидов.  
 Знаком полукруга под уровнем  
 обозначены состояния, с которых  
 наблюдалось свечение



# Levels in band scheme

P. Dorenbos / Journal of Luminescence 135 (2013) 93–104



$\text{LaF}_3$

Sm, Eu, Tm, Yb - can be divalent  
Possibility of “anomalous” luminescence for divalent lanthanides

# Anomalous luminescence of $\text{Eu}^{2+}$ and $\text{Yb}^{2+}$ in inorganic compounds

P Dorenbos

Interfaculty Reactor Institute, Delft University of Technology, Mekelweg 15, 2629 JB Delft, The Netherlands

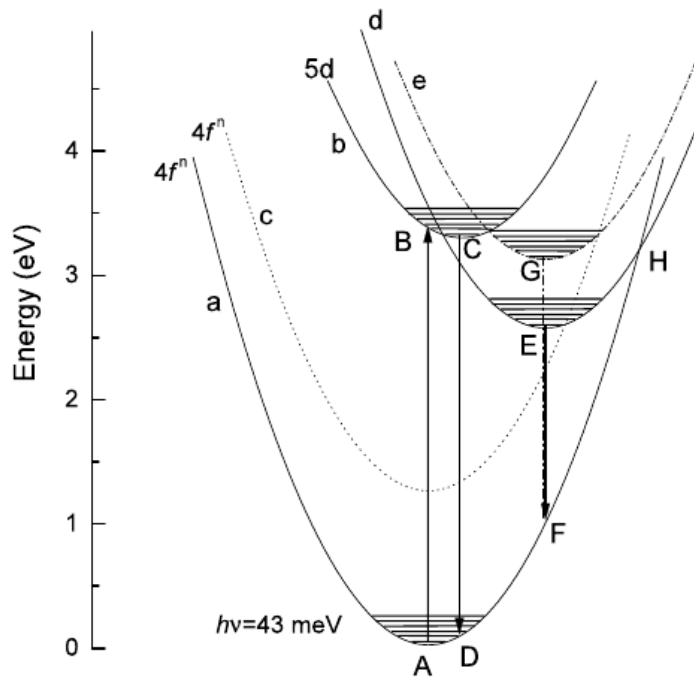


Figure 2. The configuration coordinate diagram illustrating normal  $df$  emission and anomalous emission. Energy values realistic for  $\text{SrF}_2$  and  $\text{BaF}_2$  were used with  $h\nu = 43 \text{ meV}$ .

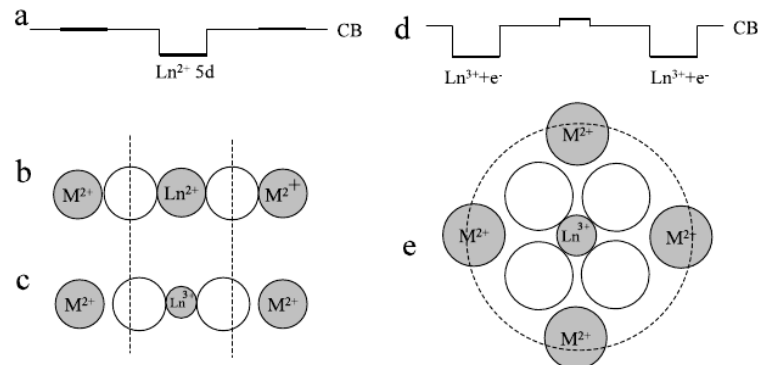
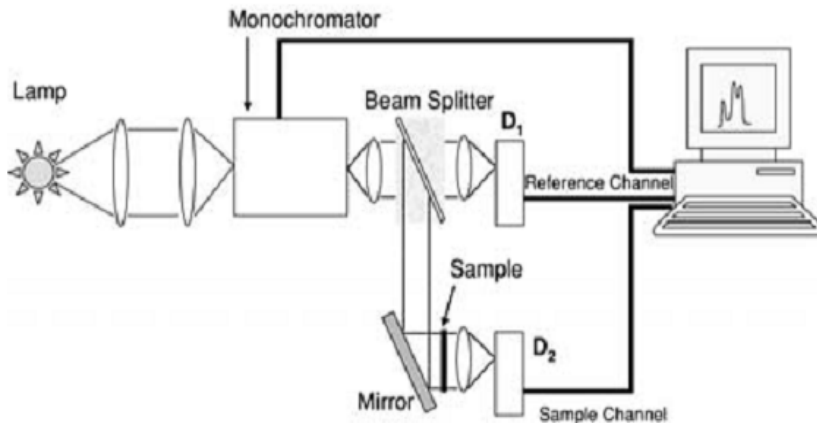
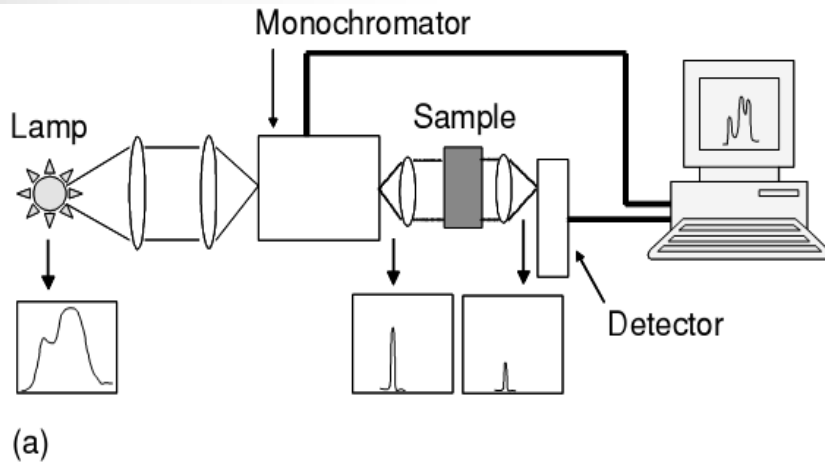


Figure 3. An illustration of the impurity-trapped exciton configuration in an  $\text{MX}_2$  type of compound. (a) The occupied  $5d$  energy level located just below the conduction band. (b) Ionic radii before the  $fd$  transition. (c) Ionic radii plus relaxation after autoionization. (d) Energy levels after autoionization and relaxation. (e) An electron orbiting around  $\text{Ln}^{3+}$  in the impurity-trapped exciton configuration.

Table 1. Properties of  $\text{Eu}^{2+}$ -doped compounds with (suspected) anomalous luminescence. Wavelengths are in nanometres and Stokes shifts  $\Delta S$  and widths  $\Gamma$  are in electron volts. Unless otherwise indicated, the width is specified at room temperature. ‘xx’ means that broadband emission is not observed down to  $\approx 10 \text{ K}$ .

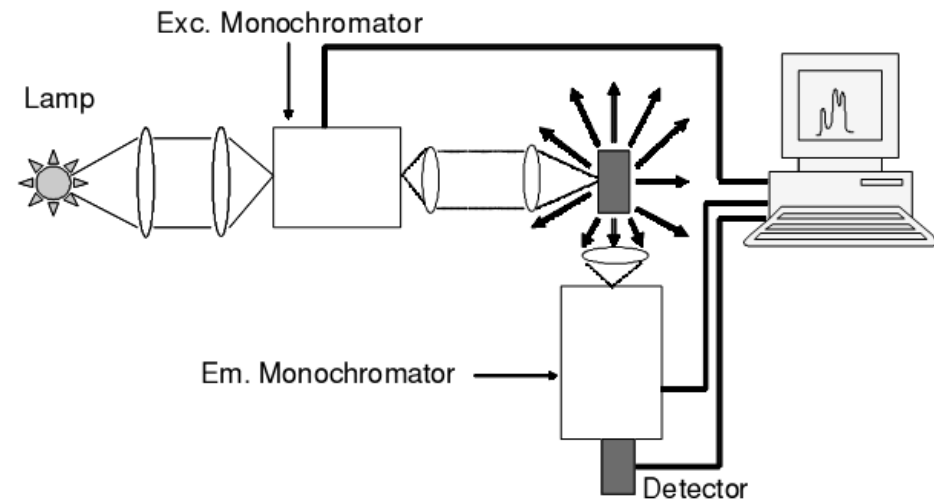
Compound	$\lambda_{\text{abs}}$	$\lambda_{\text{em}}^{\text{df}}$	$\lambda_{\text{em}}^{\text{anom}}$	$\Delta S^{\text{anom}}$	$\Gamma_{\text{Eu}}^{\text{anom}}$
$\text{BaF}_2$	382	403	590	1.14	0.51 (77 K)
$\text{BaLiF}_3$	333		410	0.70	0.37 (270 K)
$\text{CdF}_2$	407		xx		
$\text{CsCaF}_3$ (300 K)	425		510	0.49	0.60
$\text{CsCaF}_3$ (77 K)	425		610	0.88	0.48 (77 K)
$\text{RbMgF}_3$ (Rb)	340	365	405	0.59	0.40
$\text{Ba}_5(\text{PO}_4)_3\text{F}$ (6h)		432	475		
$\text{Sr}_2\text{LiSiO}_4\text{F}$	400		533	0.77	0.49
$\text{Ba}_2\text{Y}(\text{BO}_3)_2\text{Cl}$		538	634		0.30 (4 K)
$\text{Cs}_2\text{SO}_4$	378		450	0.52	0.66

# Измерение спектров



$$T = I/I_0$$

$$OD = \log(I_0/I),$$



$$I_{em} = \eta(I_0 - I)$$

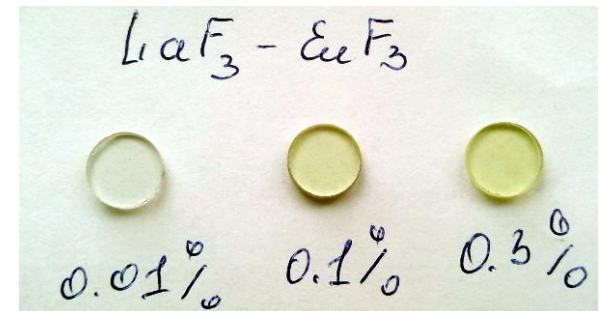
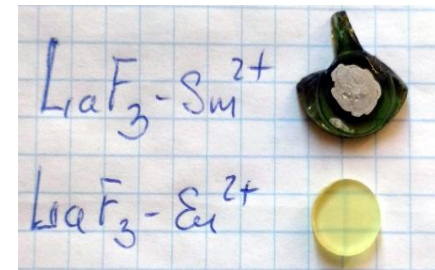
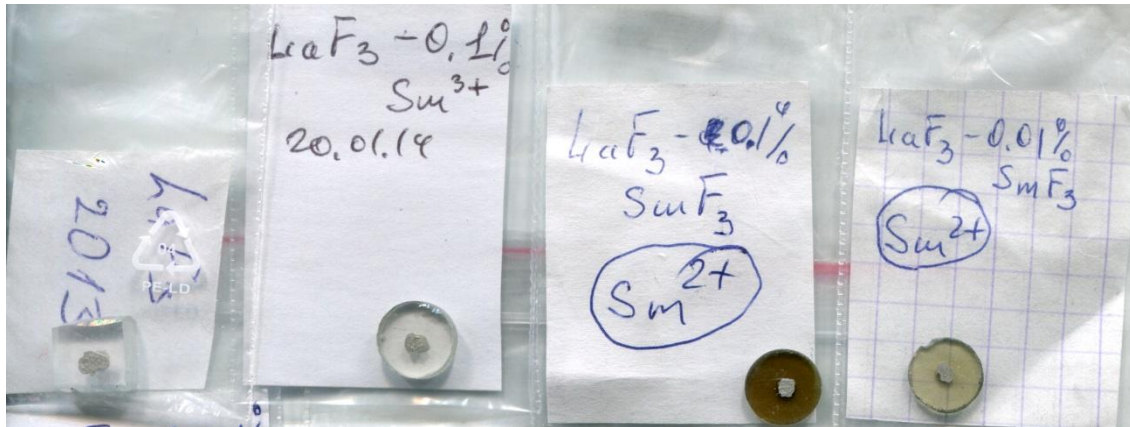
$$(I_{em}) = k_g \times \eta \times I_0(1 - 10^{-(OD)})$$

$$(I_{em}) \cong k_g \times \eta \times I_0 \times (OD)$$

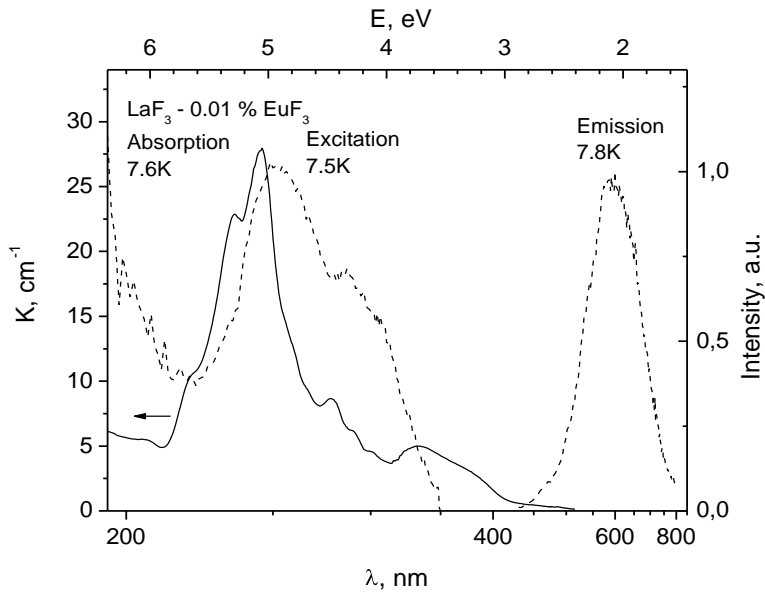
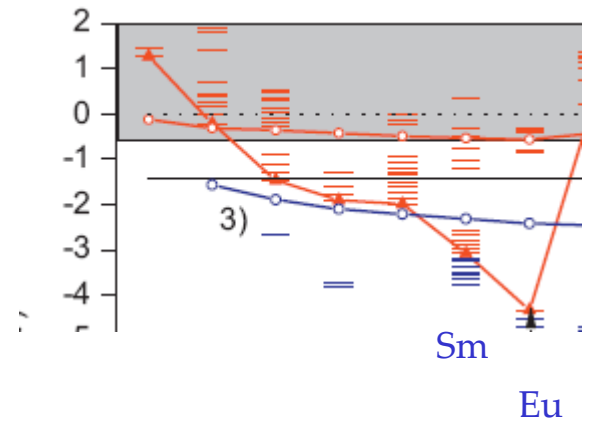


# Experimental

- The crystals were grown by Stockbarger method in graphite crucible in vacuum. A few percent of  $\text{CdF}_2$  was added into raw materials for purification from oxygen impurity during growth. Impurity  $\text{RE}_2\text{F}_7$  was added into  $\text{LaF}_3$  powder in concentration of 0.01, 0.1 and 0.3 mol. % .
- They were grown crystals containing only  $\text{Sm}^{3+}$ , and crystals in which a substantial proportion of the trivalent samarium converted to divalent form. Color of crystals  $\text{LaF}_3\text{-Sm}^{2+}$  varied from light to dark green with increasing concentration samarium. Crystals  $\text{LaF}_3\text{-Sm}^{3+}$  are colorless.
- The crystals are polished to optical measurements . For conductivity measurements on the polished surface applied electrodes ( conductive silver adhesive " kontaktol ")



# Spectra of $\text{Eu}^{2+}$

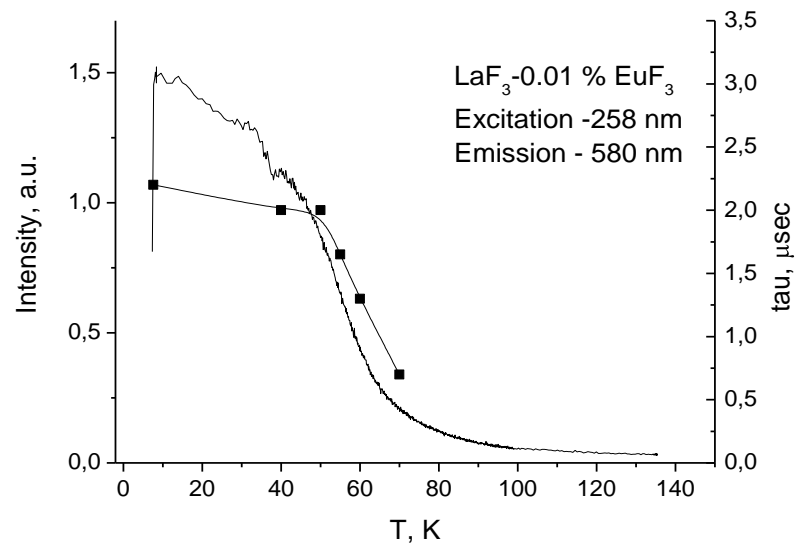
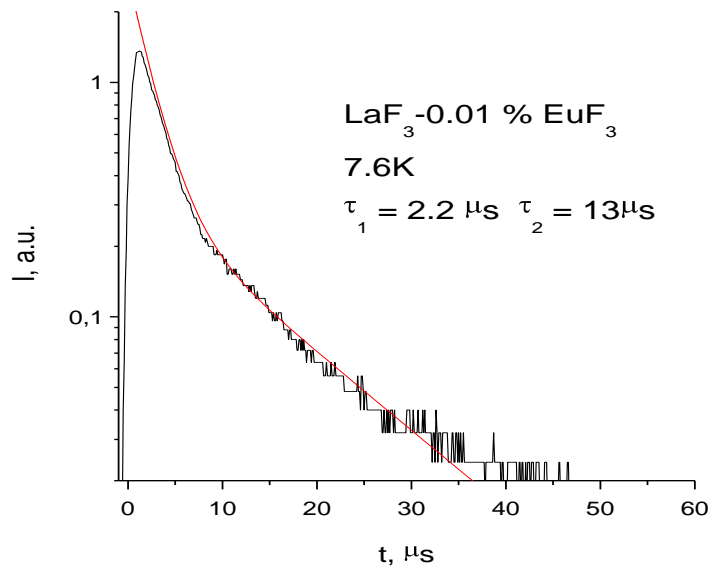


Luminescence with large Stokes shift  
Excited within f-d region.

Anomalous  $\text{Eu}^{2+}$  luminescence

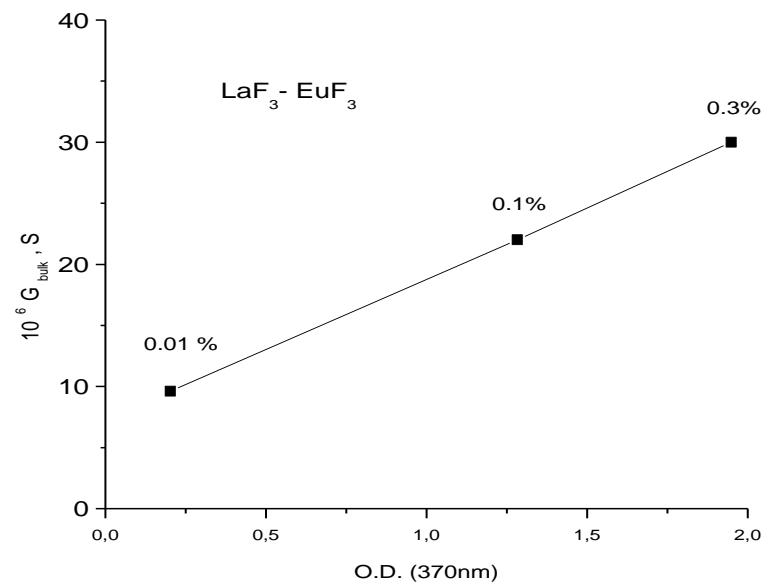
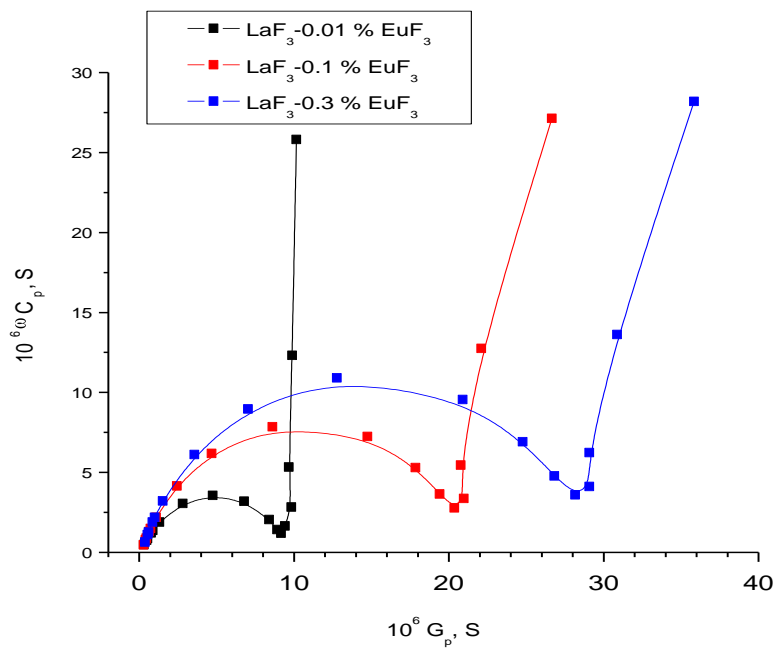
Absorption, emission and excitation spectra of  $\text{LaF}_3\text{-}0.01\% \text{Eu}^{2+}$

# Luminescence of $\text{LaF}_3\text{-Eu}^{2+}$

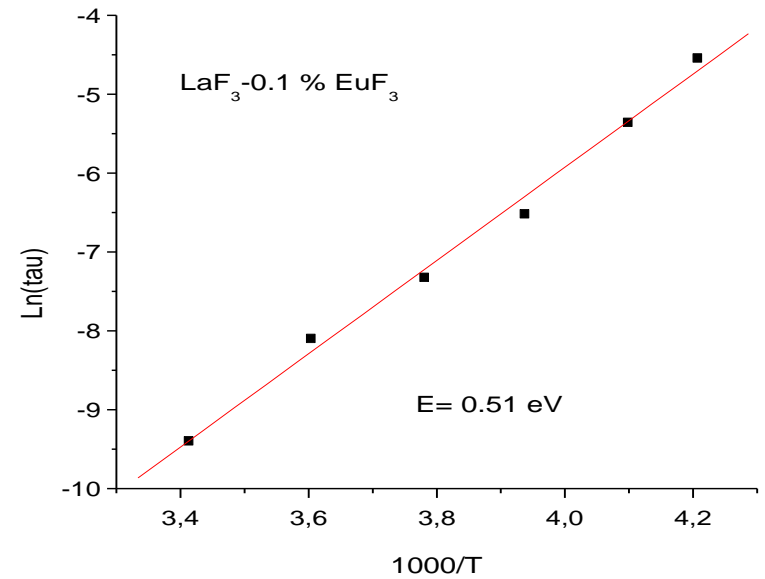
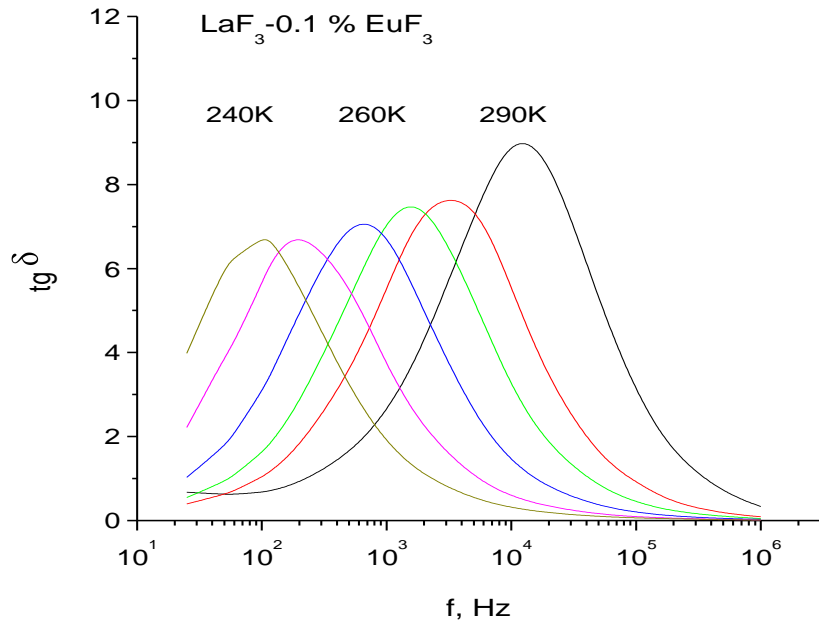


Temperature dependence of emission and decay times of  $\text{LaF}_3\text{-0.01 \% Eu}^{2+}$

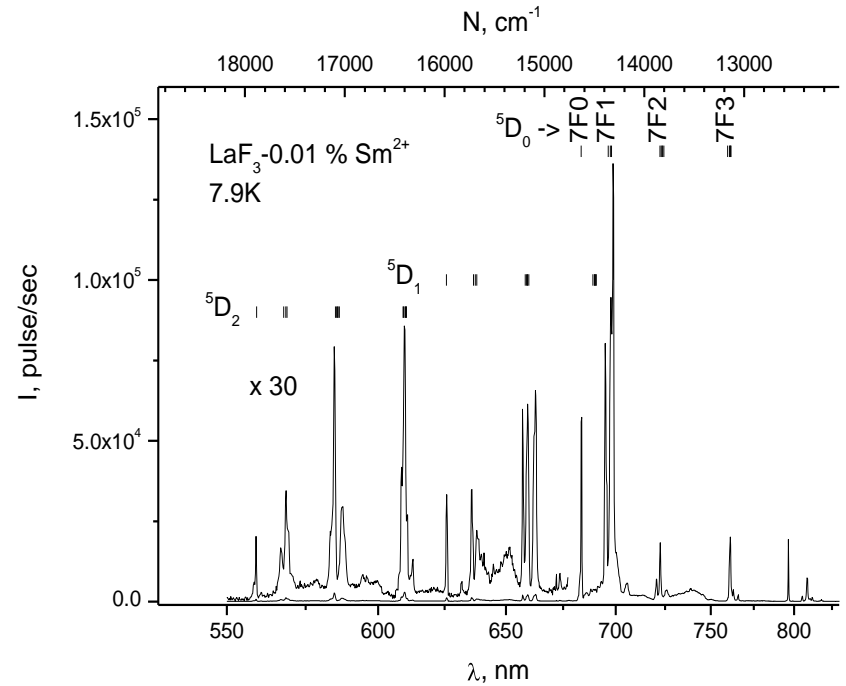
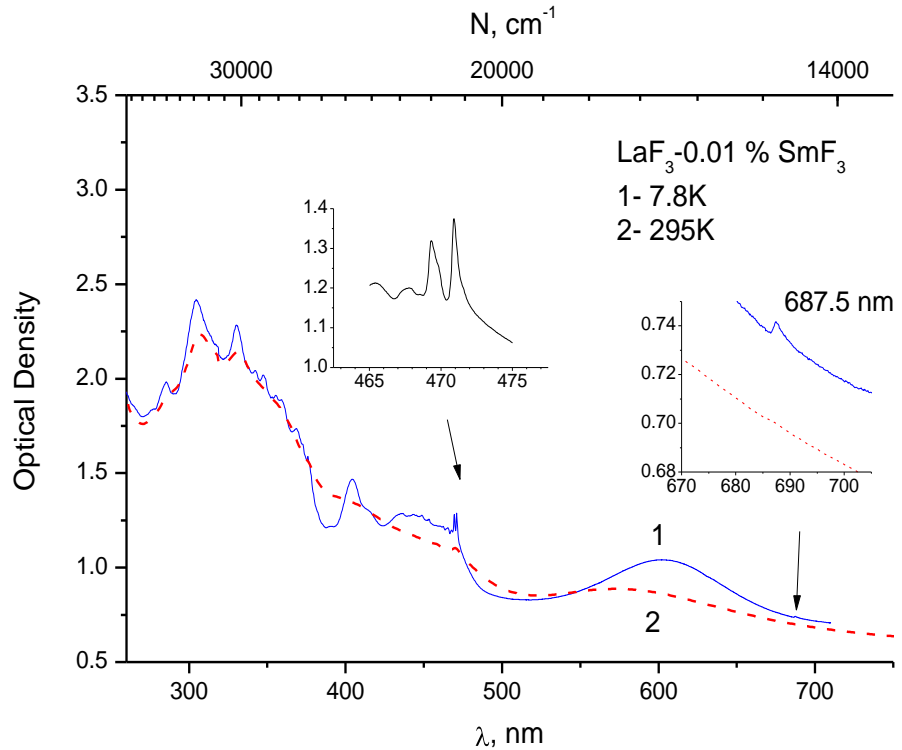
# LaF<sub>3</sub>-Eu<sup>2+</sup> conductivity



# LaF<sub>3</sub>-Eu<sup>2+</sup> dielectric loss



# Spectra of $\text{Sm}^{2+}$

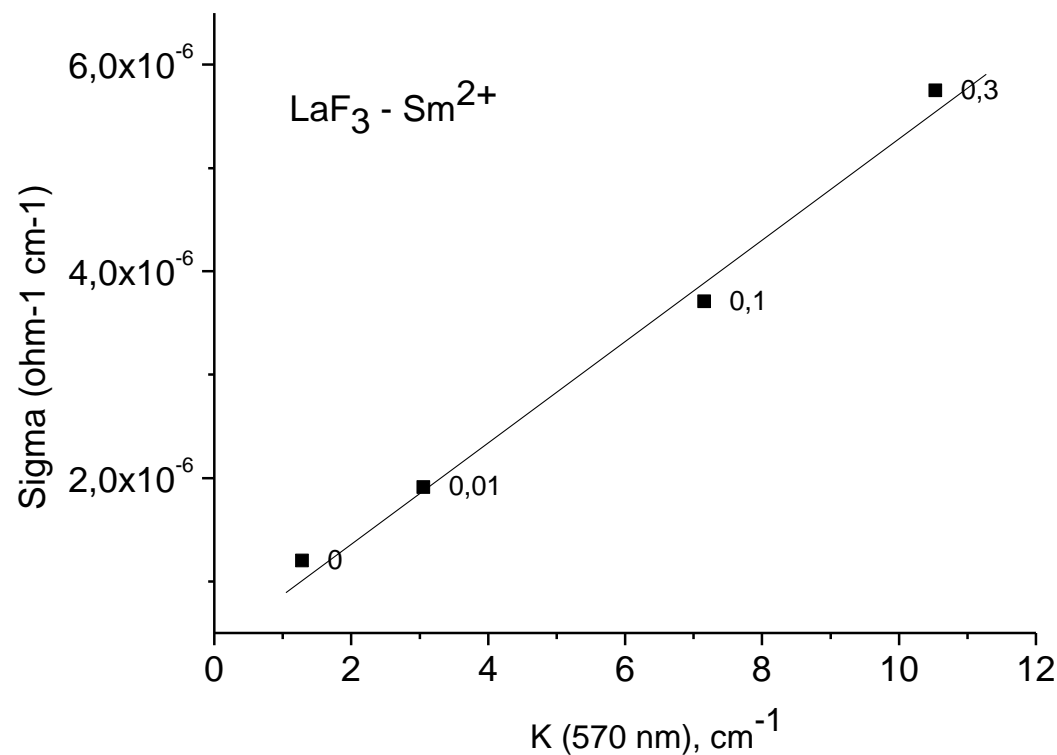


Absorption and emission spectra of  $\text{Sm}^{2+}$  in  $\text{LaF}_3$ - 0.01 %  $\text{SmF}_3$ .

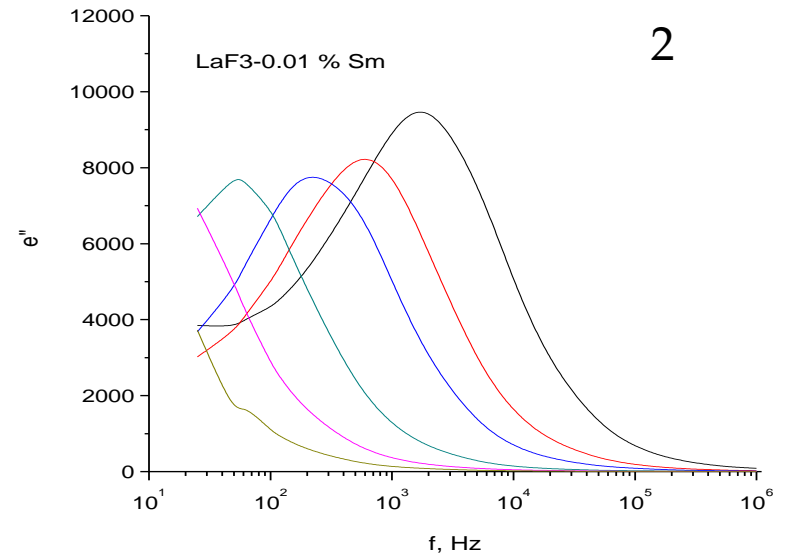
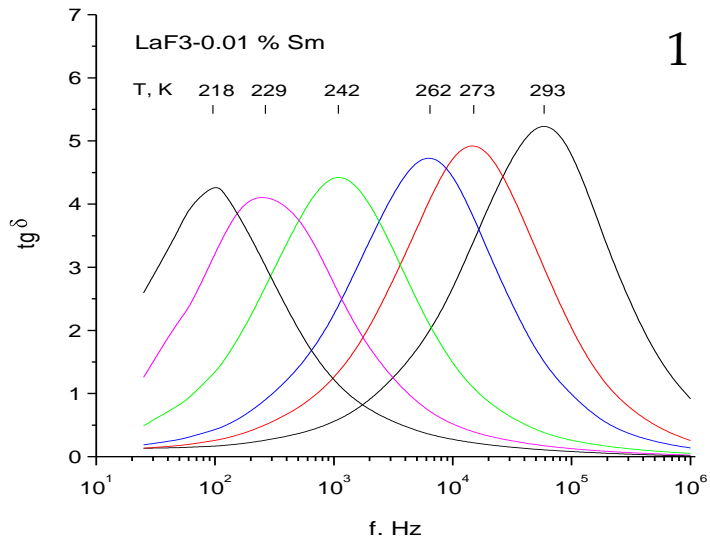
# LaF<sub>3</sub>-Sm<sup>2+</sup> conductivity

Conductivity of LaF<sub>3</sub>-Sm<sup>2+</sup> samples against of absorption coefficient of band at 570 nm at room temperature. Concentration of doped SmF<sub>3</sub> are shown near experimental points.

кристалл	10 <sup>6</sup> σ (ом <sup>-1</sup> см <sup>-1</sup> )
LaF <sub>3</sub> (L13)	0,37
LaF <sub>3</sub> -0.1 % Sm <sup>3+</sup>	1,20
LaF <sub>3</sub> -0.3 % Sm <sup>2+</sup>	5,75
LaF <sub>3</sub> -0.1 % Sm <sup>2+</sup>	3,71
LaF <sub>3</sub> -0.01 % Sm <sup>2+</sup>	1,91
LaF <sub>3</sub> -0.1% EuF <sub>3</sub> (Eu <sup>2+</sup> )	3,33



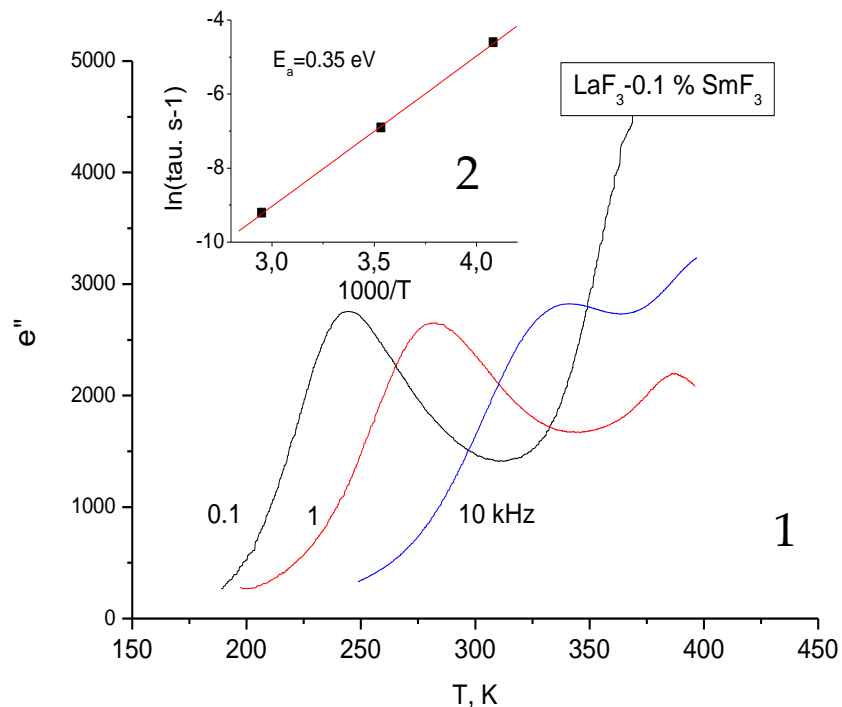
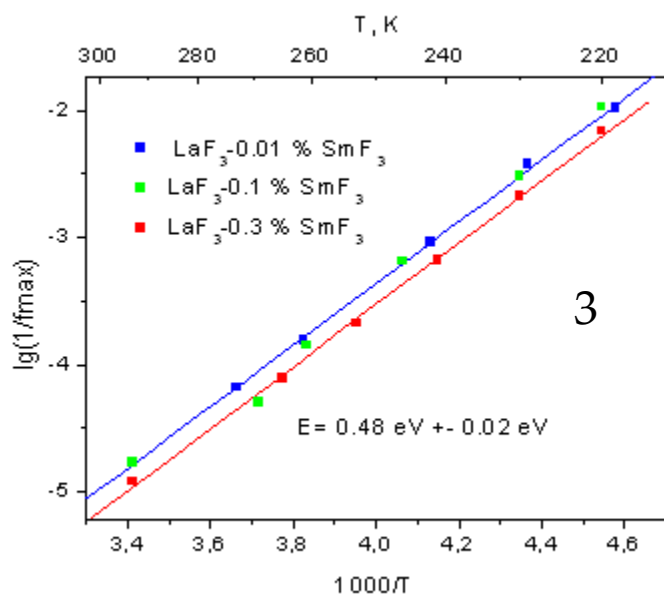
# Dielectric losses $\text{LaF}_3\text{-Sm}^{2+}$



Temperature dependence of  $\text{tg}(\delta)$  (1) and  $e''$  (2) of  $\text{LaF}_3\text{-Sm}^{2+}$

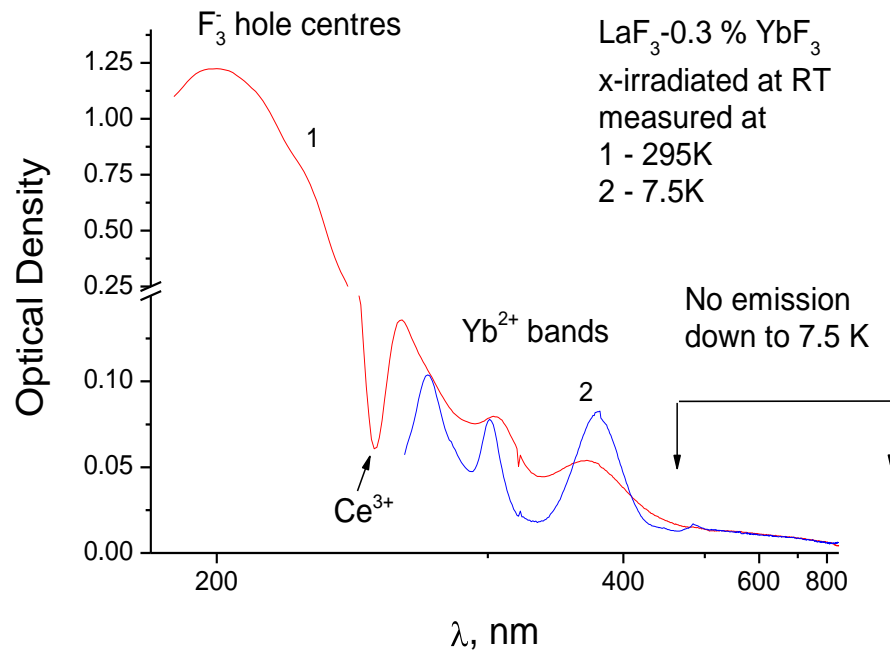


# Dielectric losses $\text{LaF}_3\text{-Sm}^{2+}$



Temperature dependence of  $\epsilon''$ (1) and Arrhenius plots (2,3) of  $\text{LaF}_3\text{-Sm}^{2+}$

# Spectra of Yb<sup>2+</sup>

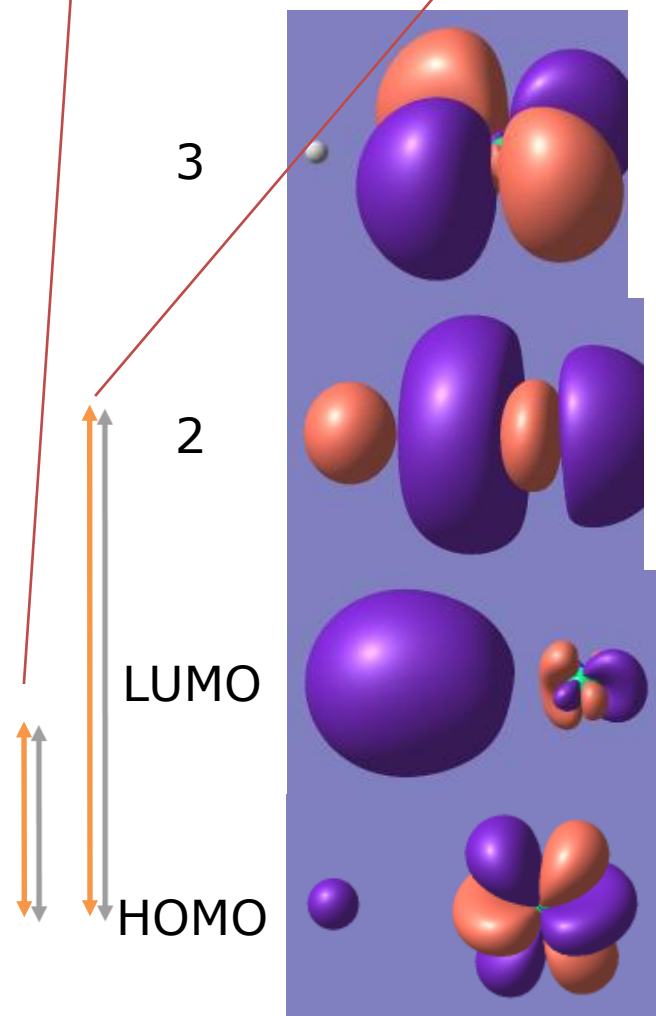
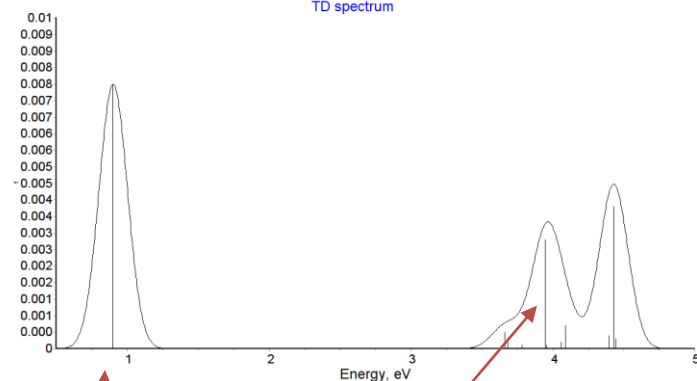
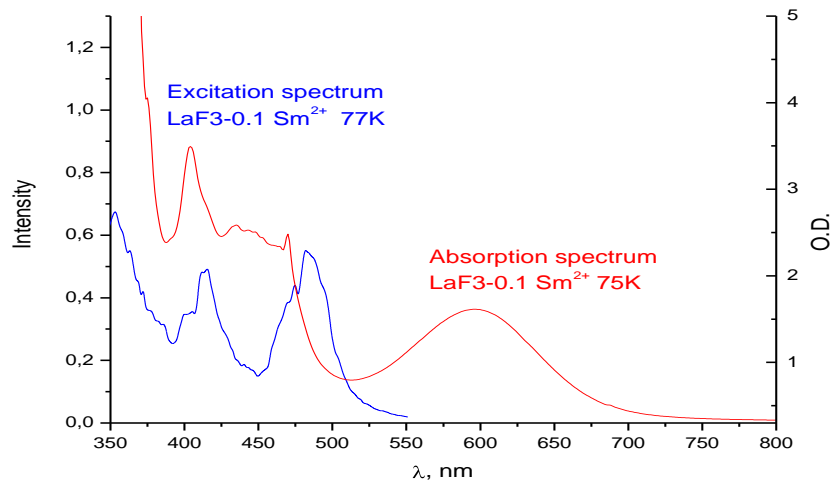
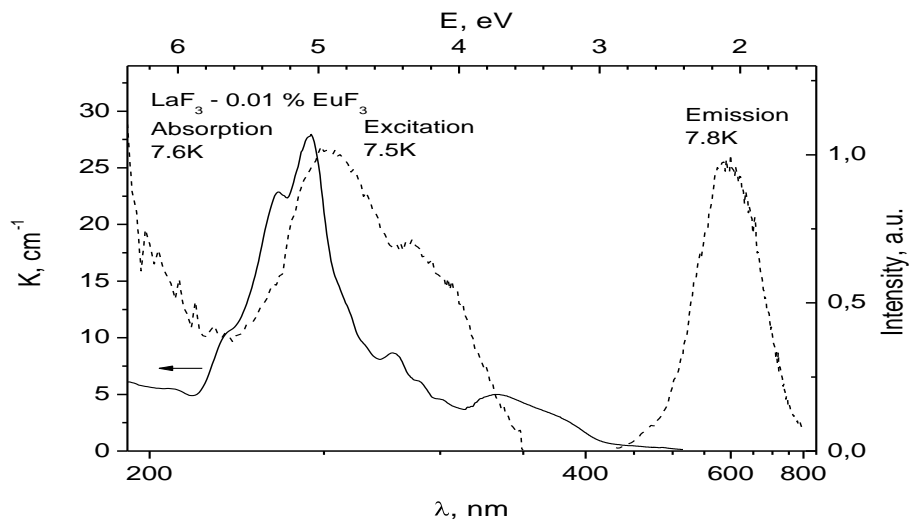


Yb<sup>2+</sup> ions were obtained by X-irradiation (Pd 20mA 40kV) of LaF<sub>3</sub>-YbF<sub>3</sub> crystal. Small part of Yb<sup>3+</sup> was reduced to Yb<sup>2+</sup> with absorption bands at 310, 370 nm. No emission of Yb<sup>2+</sup> was observed.

Absorption spectra of LaF<sub>3</sub>-0.3% YbF<sub>3</sub> created by X-irradiation at 295K

# Model

Molecule  
 $\text{Sm}^{3+} - \text{H}$



# Conclusion

- Finally, a set of studies allow us to establish the nature of the charge compensator of divalent rare earth  $\text{Sm}^{2+}$  and  $\text{Eu}^{2+}$  ions and explain the details of the optical and dielectric properties of these centers in the lanthanum fluoride crystals